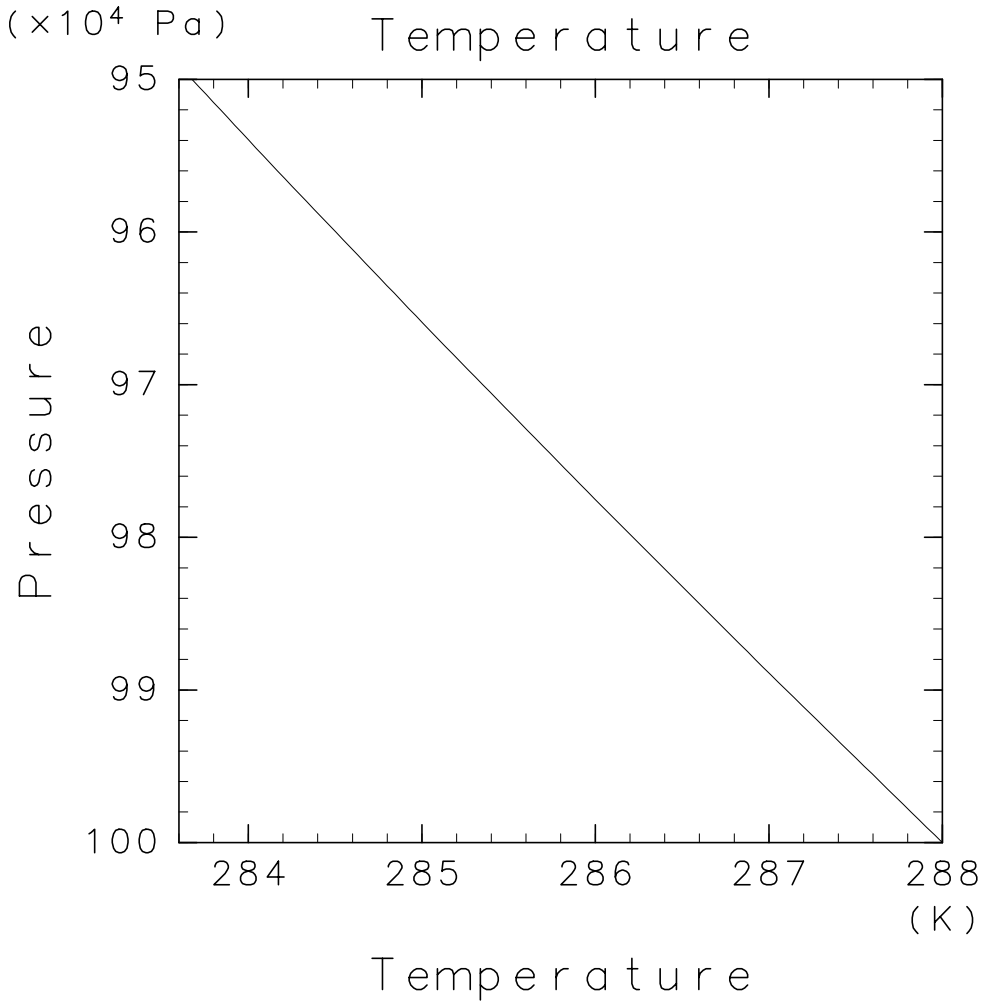
(×10⁴ Pa)
Temperature
95
96
97
98
99
100
Pressure
284
285
286
287
288
(K)
Temperature

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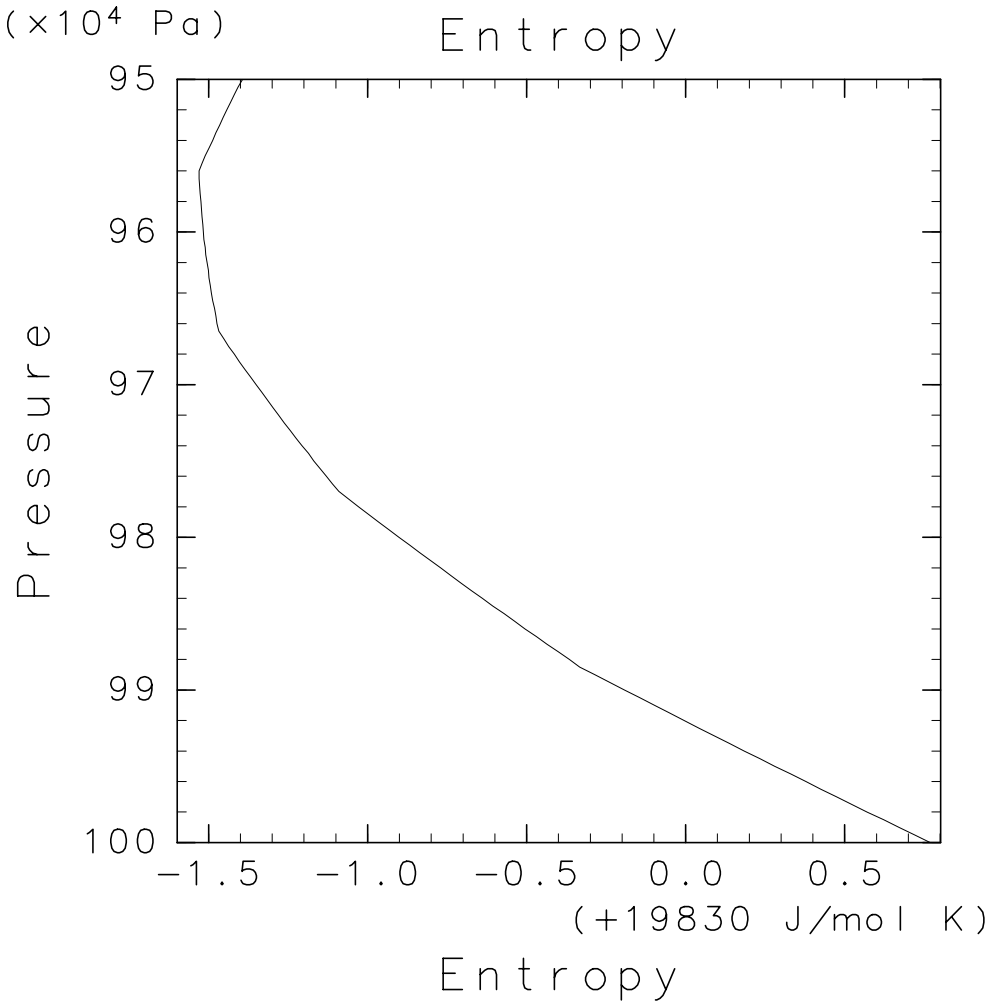
(×10⁴ Pa)
Entropy
95
96
97
98
99
100
Pressure
−1.5
−1.0
−0.5
0.0
0.5
(+19830 J/mol K)
Entropy

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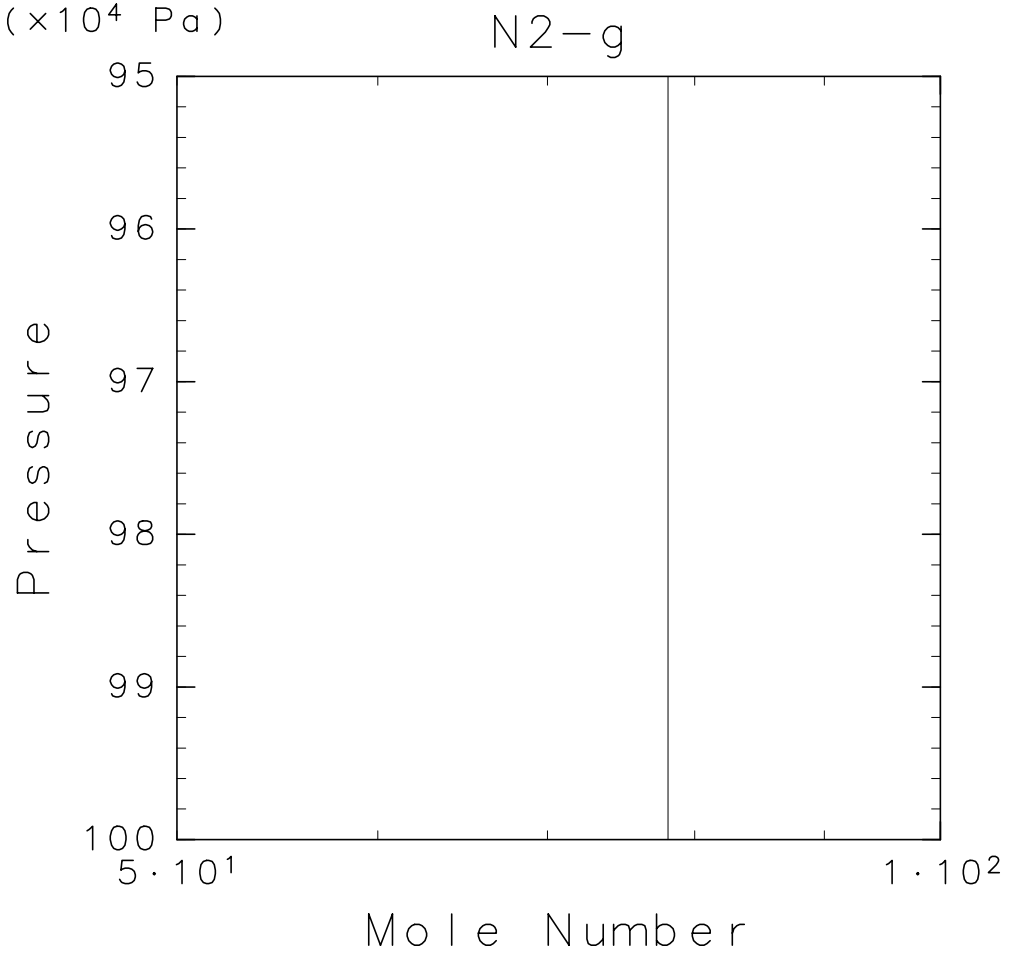

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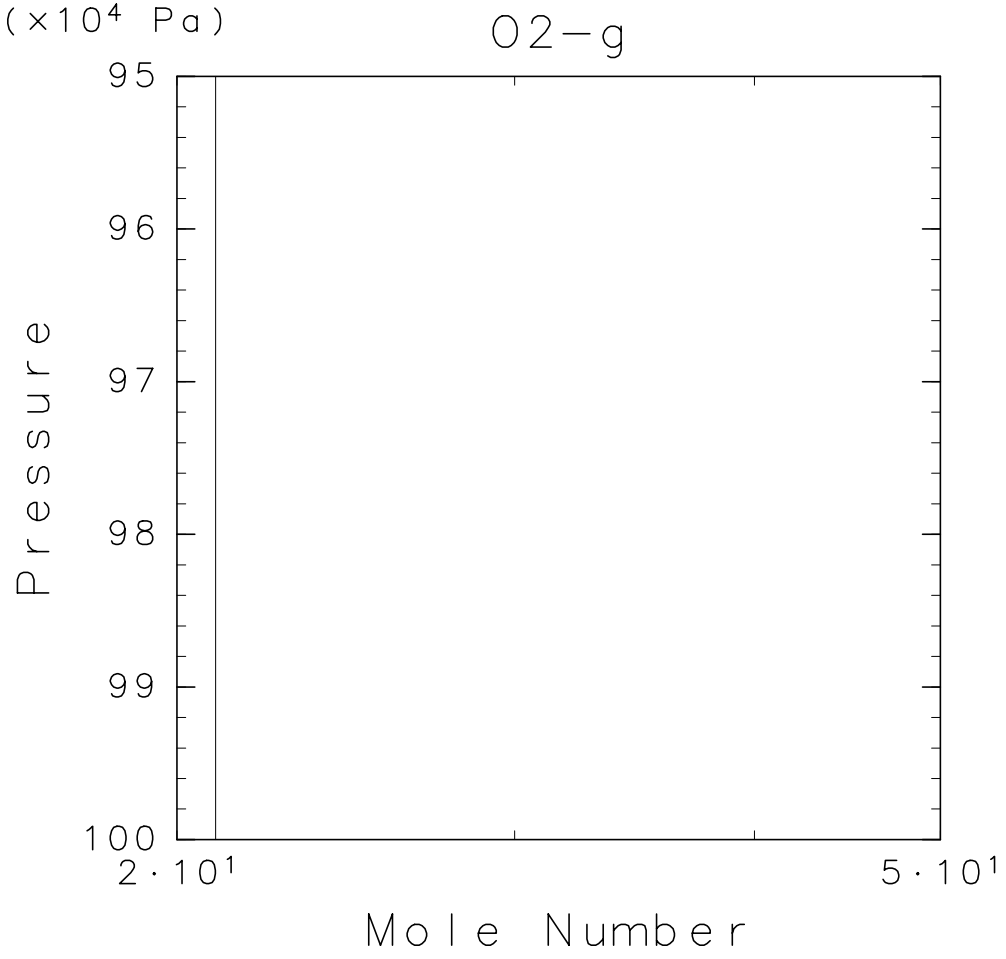

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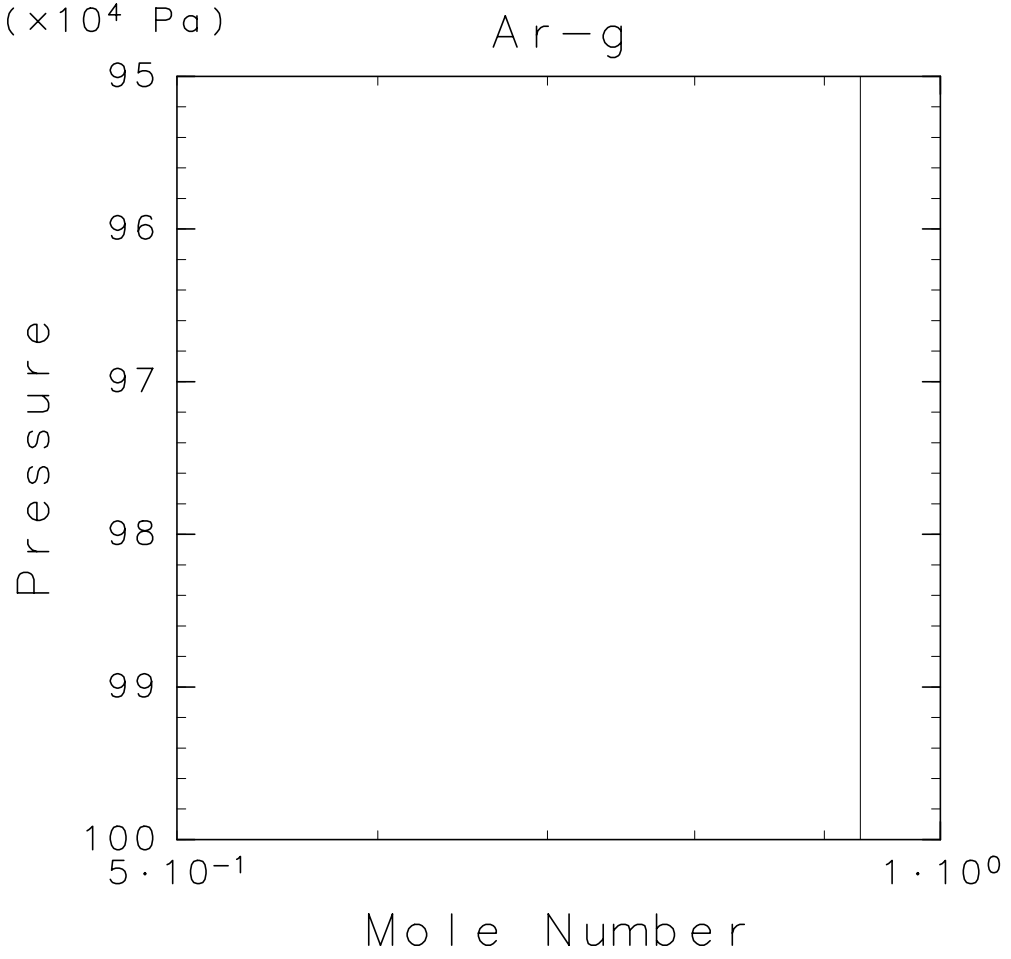

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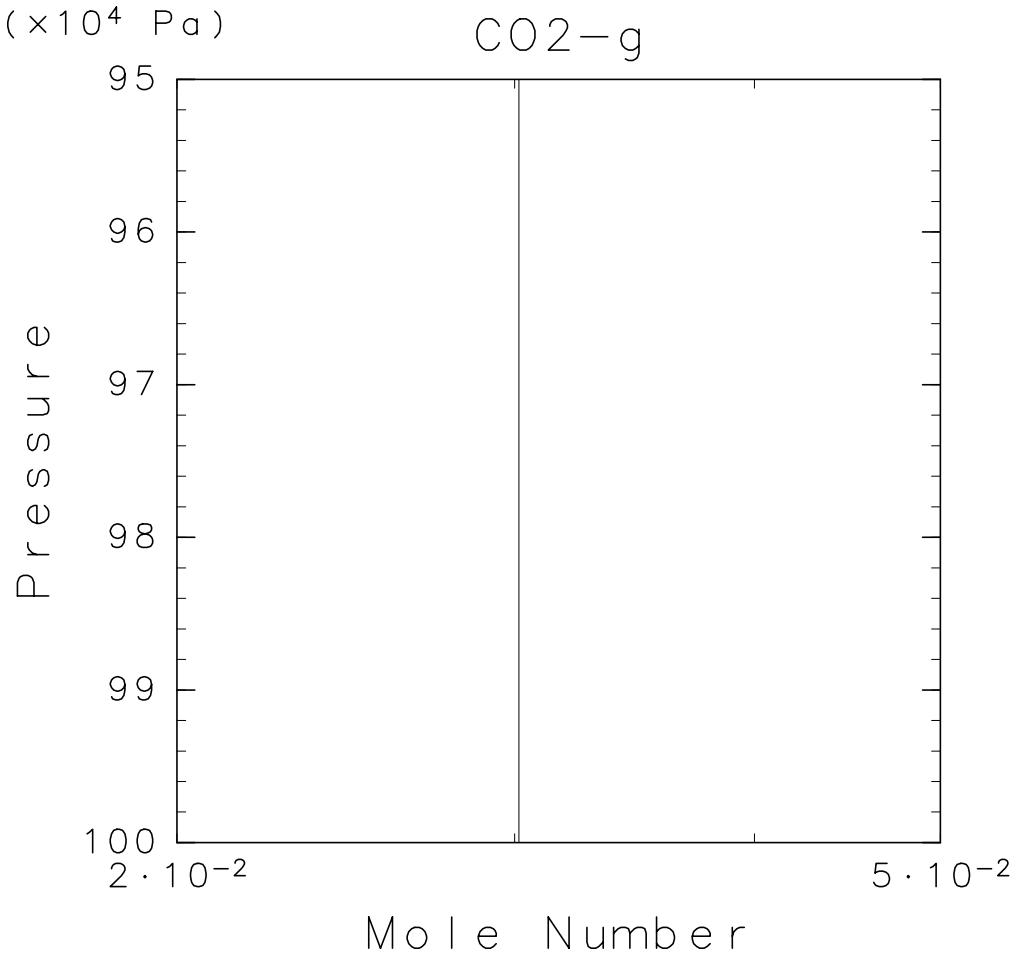

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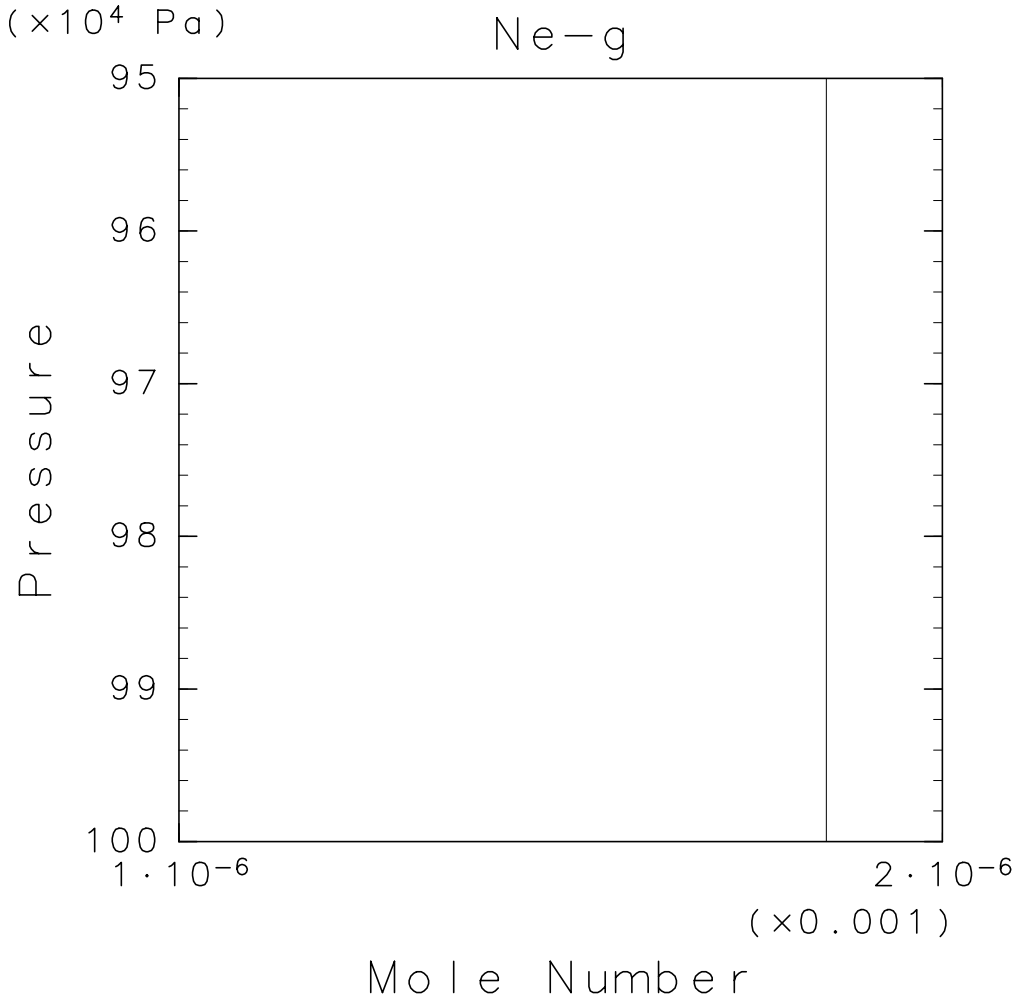
$(\times 10^{4}\ \mathrm{Pa})$

$(\times 10^{-20})$


(×10⁴ Pa)
Ne−g
95
96
97
98
99
100
Pressure
1·10⁻⁶
2·10⁻⁶
(×0.001)
Mole Number

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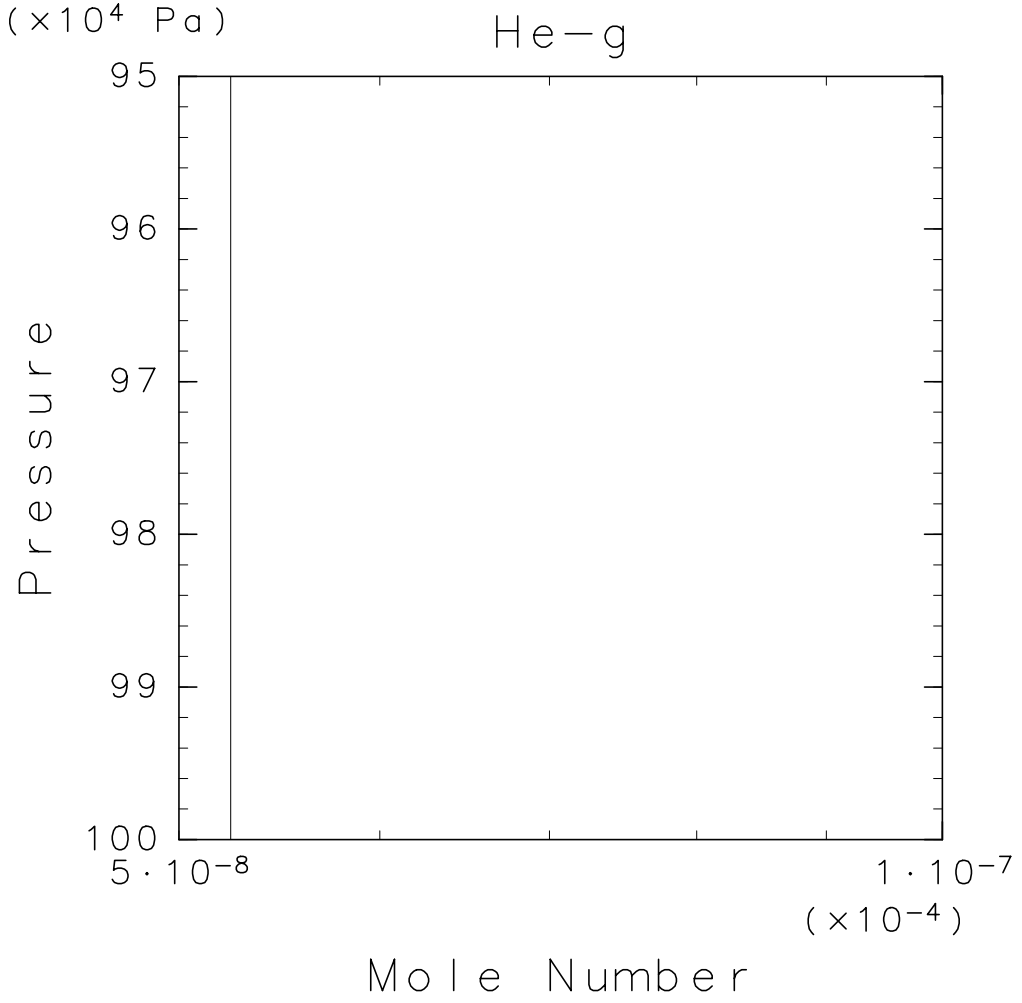
(×10⁴ Pa)
He−g
95
96
97
98
99
100
Pressure
5·10⁻⁸
1·10⁻⁷
(×10⁻⁴)
Mole Number

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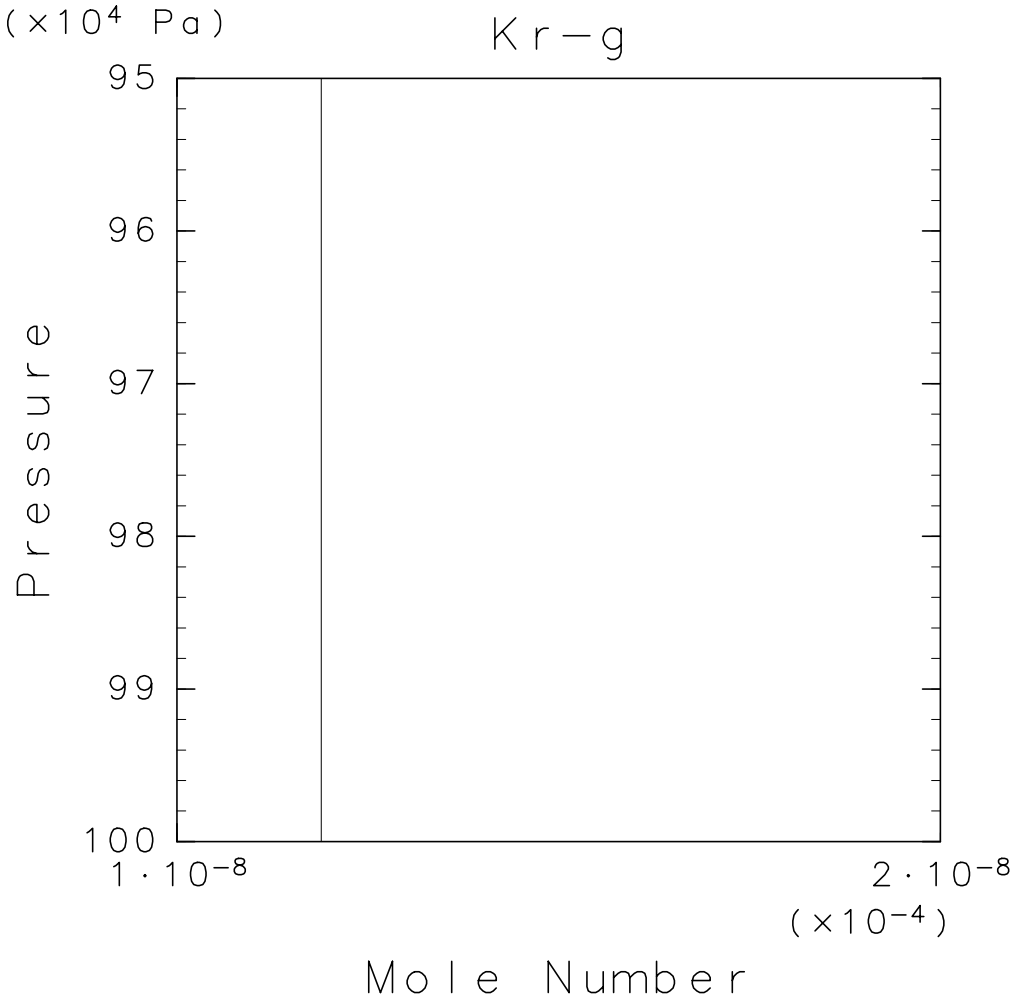
$(\times 10^{4}\ \mathrm{Pa})$

$(\times 10^{-20})$


(×10⁴ Pa)
Kr−g
95
96
97
98
99
100
Pressure
1·10⁻⁸
2·10⁻⁸
(×10⁻⁴)
Mole Number

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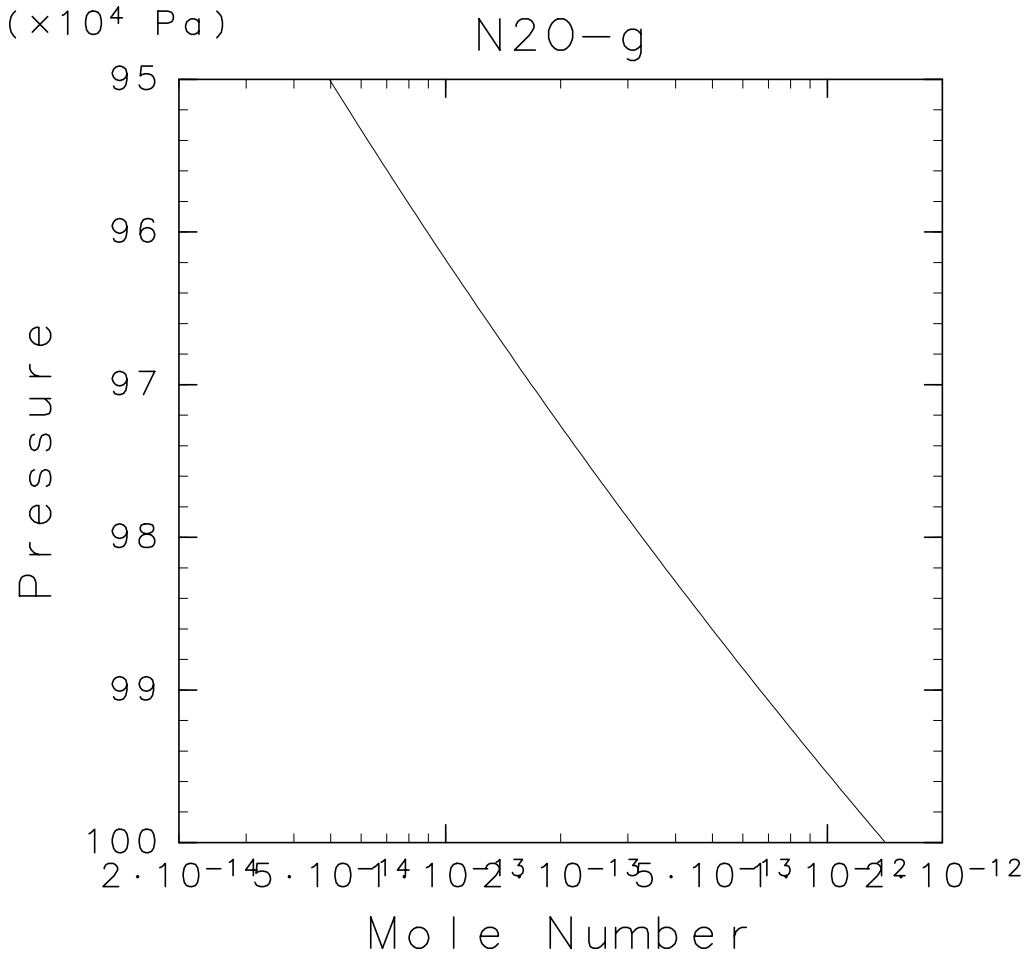
(×10⁴ Pa)
N2O−g
95
96
97
98
99
100
Pressure
2·10⁻¹⁴
5·10⁻¹⁴
10⁻¹³
2·10⁻¹³
5·10⁻¹³
10⁻¹²
2·10⁻¹²
10⁻¹²
Mole Number

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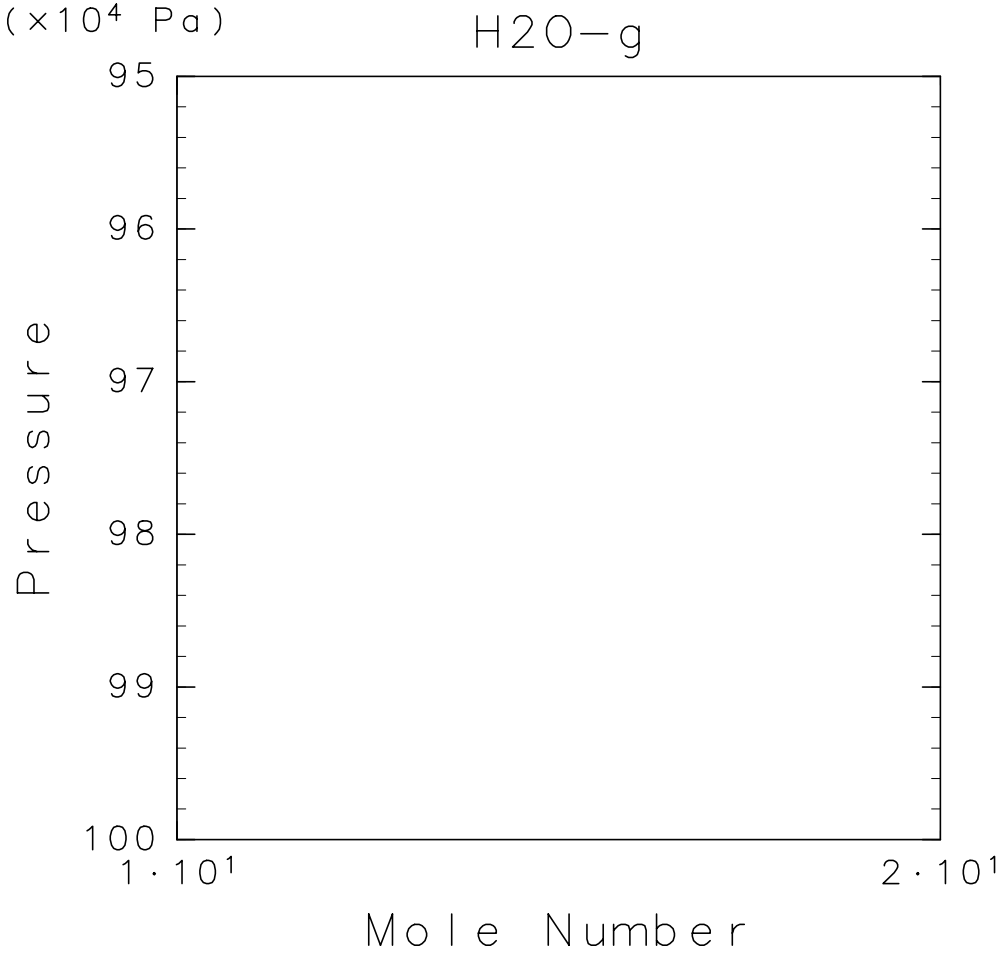
$(\times 10^{4}\ \mathrm{Pa})$

$(\times 10^{-20})$



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